

**Reducing Risk and Mitigating Cost  
by Understanding the Source of H<sub>2</sub>S.  
Examples from Texas**

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No operator wants to have a sour gas problem in their well. Even worse, “Surprise, we detected 10, 1000 or > 10,000 ppm H<sub>2</sub>S in your well!”

The presence of H<sub>2</sub>S in the production stream is a well-known problem from a safety and commercial perspective. Understanding the H<sub>2</sub>S source is critical to estimating potential production quantities over time as well as identification of H<sub>2</sub>S trends. Source identification allows critical cost and associated risk reduction decisions such as evaluating chemical treatment vs sour facility to be made early in the life of the field.

Sulfur is incorporated into organic matter shortly after deposition, generally within the first few meters of the subsurface, via reduction of seawater sulfate by bacteria. If there are abundant metals available, the sulfur will preferentially form metal sulfides. There are three main sources of H<sub>2</sub>S in petroleum

Bacterial reduction of sulfate to H<sub>2</sub>S through dissimilatory reactions. The biological creation of H<sub>2</sub>S requires conditions in the subsurface compatible with bacterial growth, the most important of which is reservoir temperatures <80°C.

Thermochemical sulfate reduction = TSR. TSR is a reaction of sulfate minerals (primarily anhydrite) and petroleum to form H<sub>2</sub>S and carbonates. This reaction is thought to begin only at subsurface temperatures > 120°C or more.

Thermal decomposition of sulfides in kerogen in clay-poor sulfur rich source rocks. The reactions are high temperature at the late stages of thermal maturation in carbonate dominated source rocks. Associated oils are high in sulfur and can provide a source for the H<sub>2</sub>S in addition to the primary kerogen.

Determining the origin of H<sub>2</sub>S found in any particular accumulation requires analysis of the H<sub>2</sub>S itself, as well as the other components of the natural gas, the water, and possibly

any produced liquids. Examples from Texas will include thermochemical  $H_2S$  that migrated from deep reservoirs, biological  $H_2S$  generated *in situ* and a combination of both types.

Once the source has been identified informed economic decisions with respect to treatment, facility types and even where to market (sour or sweet) can be made early in field development.

